

# Pulsed Electrical Field Impact on Cyanide Biodegradation by *Pseudomonas fluorescens*

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The impact of the pulsed electrical field on the cyanide biodegradation process was investigated in this work. In the experiment, *Pseudomonas fluorescens* bacteria was treated by the pulsed electrical field for 15 mins ( $f = 100$  Hz, impulse duration is 1 ms) before adding to the solution with  $\text{Na}[\text{Ag}(\text{CN})_2]$  complex, and the appropriate kinetics was described [1]. During the cyanide biodegradation process, cyanide blocks the respiratory centers (RCs) of bacteria, but simultaneously bacteria degrade cyanide using the respiratory mechanism [2]. Theoretical analysis of the cyanide biodegradation kinetics was carried out in [3].

The purpose of this work is to introduce a phenomenological model (1-2) that explains the cyanide biodegradation process in [1], and to describe the impact of pulsed electrical field on respiratory parameters of bacteria.

$$\frac{dn}{dt} = -(\gamma_0 + \gamma_1 C)n + (g_0 + g_1 n)(1 - n) - a(1 - C) \quad (1)$$

$$\frac{dC}{dt} = -\alpha n \frac{C}{C + C_m} \quad (2)$$

where  $n$  is a relative number of active RCs that can degrade cyanide,  $C$  is a cyanide concentration in the solution,  $\gamma(C) = \gamma_0 + \gamma_1 C$  is the rate of RC deactivation for low cyanide concentrations,  $\gamma_0$  and  $\gamma_1$  are constants.  $g(n) = g_0 + g_1 n$  is the rate of RC activation,  $g_0$  and  $g_1$  are constants.  $\alpha$  is the maximum rate of the cyanide destruction,  $C_m$  is the Michaelis constant,  $a$  is the rate of RCs deactivation caused by by-product generation in the solution. Note that the system (1-2) is already normalized.

During the analysis of the dependence of absorbed oxygen on cyanide concentration from [4], we identified the following relations:  $\alpha/C_m = g_1 + g_0/D^2 - \frac{\gamma_1}{AB}$  and  $C_m = \frac{2}{B} (g_1 + g_0/D^2 - \frac{\gamma_1}{AB}) / (2\frac{\gamma_1}{AB} - 2\frac{g_0 A}{D^3} - g_1 - \frac{g_0}{D^2})$ , where  $A$ ,  $B$  and  $D$  are constants. Thus parameters responsible for the rate of cyanide biodegradation are dependent on the parameters related to the respiratory activity of bacteria. In addition, we found that  $\gamma_0 = g_1 - g_1 D - g_0 + g_0/D$ . For other parameters, we identified the dependencies on the voltage of the pulsed electrical field (Fig 1).  $g_1$  and  $\gamma_1$  have linear dependence on voltage. Parameter  $a$  is not dependent on voltage. Also, we applied the aforementioned model and results to the cyanide biodegradation experiment in [4] after the re-normalization.

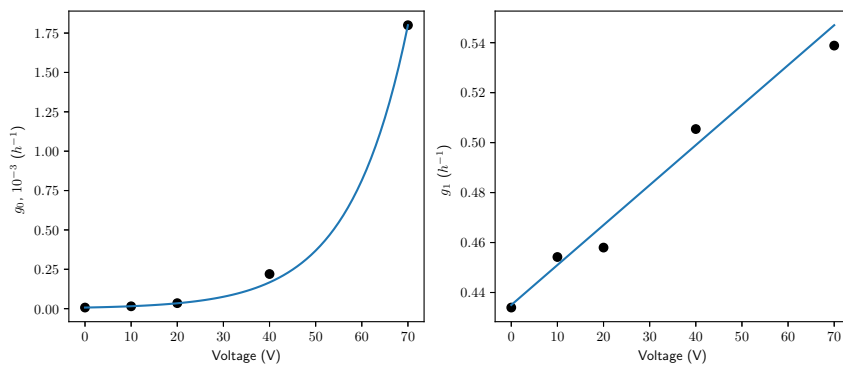


Figure 1: Dependencies of the system parameters on the voltage of pulsed electrical field

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- [4] Yakubenko L.N., Podolska V.I., Vember V.E., Karamushka V.I. The influence of transition metal cyanide complexes on the electrosurface properties and energy parameters of bacterial cells, *Colloids and Surfaces A: Physicochemical and Engineering Aspects*. **104**, 11 (1995)